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Use the Table 2.1 to answer the following questions. 6.

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Practice Problems In

your notebook, answer

the following SECTION

7.1 IONS For each

element below, state

(i) the number of

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valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most stable ion. a. Ba 2.

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structure of carbon  
dioxide is one carbon  
atom with two oxygen  
atoms on opposite  
sides of it. \_\_\_\_ 13.

Covalent bonds exist  
when combining atoms  
give up or accept  
electrons. \_\_\_\_ 14. A

molecule contains two  
atoms. Part C Matching  
Match each description  
in Column B to the  
correct term in Column



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A.

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your notebook, answer  
the following SECTION  
71 IONS For each  
element below, state  
(i) the number of  
valence electrons in

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the atom, (ii) the  
electron dot structure,  
and (iii) the chemical  
symbol(s) for the most

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the space provided.  
Date Class 11. List the  
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five traditional branches of chemistry and explain what each involves. 12. Discuss the role of chemistry in one of these areas: Materials, Energy, Medicine and Biotechnology, Agriculture, the Environment, or the Universe. 13. looks unhealthy and that, perhaps, the grass is dying.



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